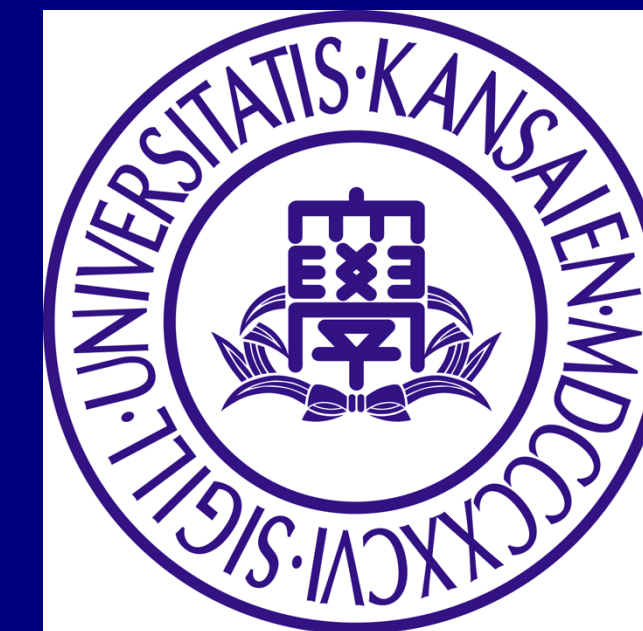


Zinc-Imidazole-Based Metal-Organic Framework Nanosheet Membrane for Gas Separation



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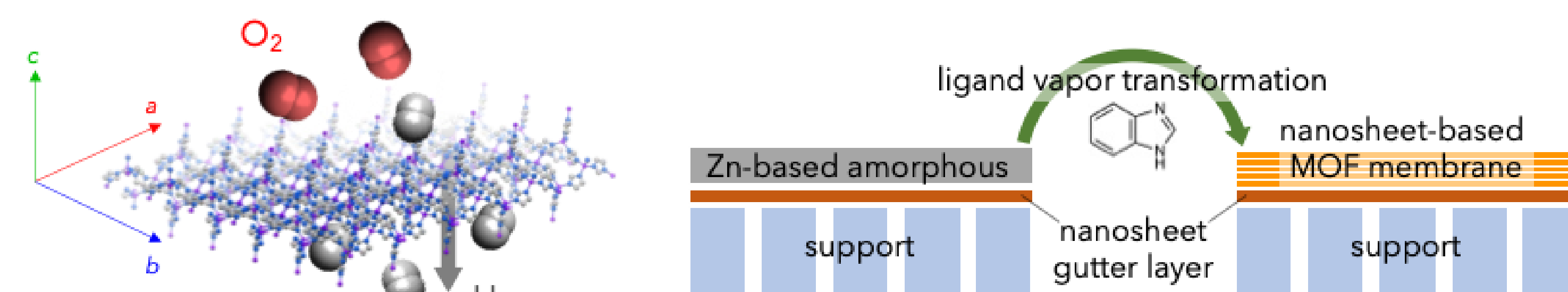
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Metal-organic framework (MOF) nanosheets are promising candidates for molecular sieve because of their structural diversity and minimized mass transfer barrier. However, design of appropriate MOF nanosheets and preparation of high-performance MOF nanosheet-based membranes, especially for gas separation, remains great challenges. Structural degradation may simultaneously occur with conventional exfoliation method, which has hindered its widespread application in high-performance membrane preparation. Even if nanosheets could be stacked, *grain boundaries would form between the nanosheets, which could be applied to liquid separation but not to gas separation.*

In this study, we developed a bottom-up method of nanosheet membrane formation in which zinc-based amorphous layer is applied on top of an intermediate gutter layer of $\text{Zn}_2(\text{benzimidazole})_4$ nanosheets fabricated by conventional interface synthesis, followed by crystallization of zinc-based layer by supplying benzimidazole vapor. **The concept of our method is to anisotropically control the crystallization of $\text{Zn}_2(\text{benzimidazole})_4$ during crystallization of the zinc-based amorphous layer by using surfactant.**

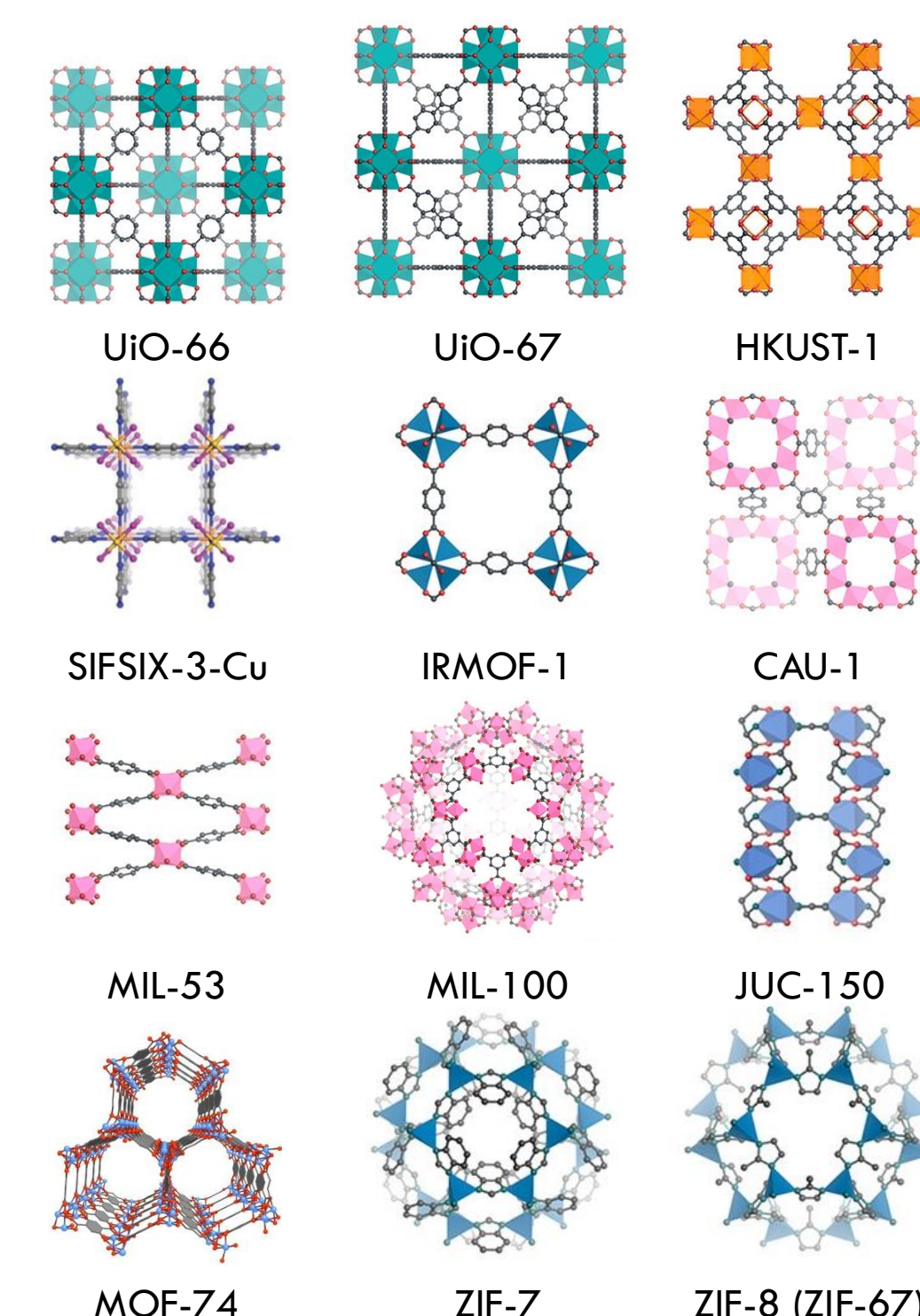


Metal Organic Frameworks (MOF)

- No template necessary
- Easy activation
- Chemical variability (metal \times linkers)
- Structural variability
- High surface area
- Framework flexibility

like graphene nanosheet

➔ MOF nanosheet membrane



Experimental

Synthesis of $\text{Zn}_2(\text{benzimidazole})_4$ nanosheet

benzimidazole in dichloromethane / zinc nitrate in aqueous solution

$\text{Zn}_2(\text{benzimidazole})_4$ nanosheet gutter layer on AAO

Zinc-based amorphous layer on gutter layer/AAO

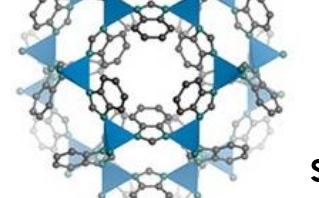
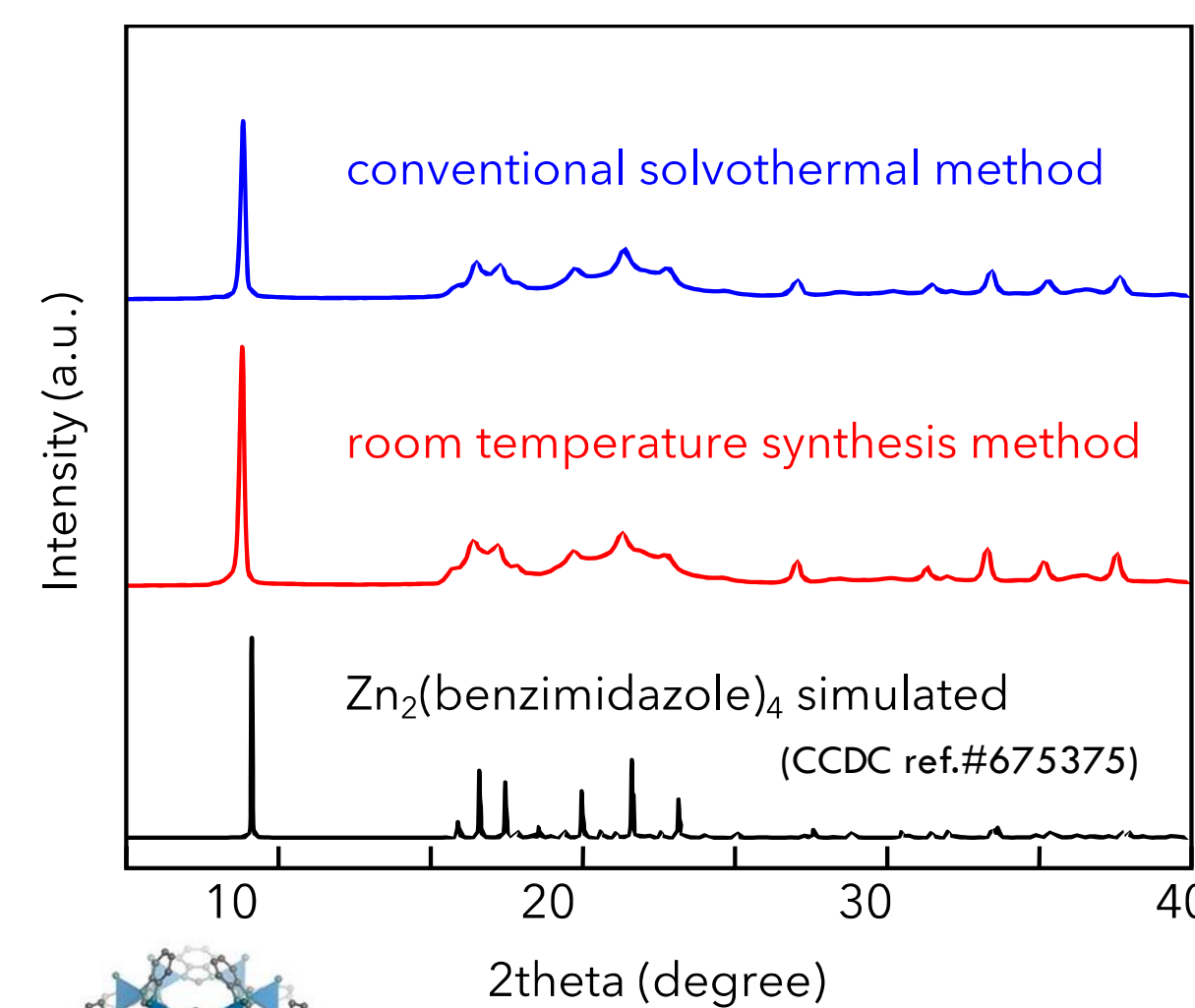
zinc nitrate, benzyltrimethylammonium chloride, 2-methoxymethanol

Conversion of zinc-based layer to $\text{Zn}_2(\text{benzimidazole})_4$ nanosheet

benzimidazole vapor deposition at 180 degC for 2 h

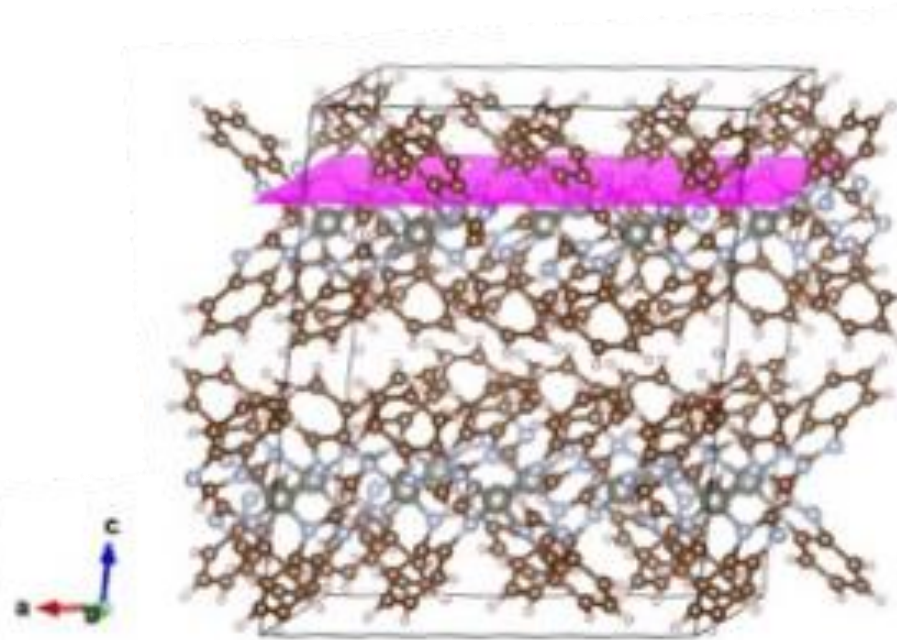
$\text{Zn}_2(\text{benzimidazole})_4$ nanosheet on gutter layer/AAO

Structure and thermal stability of bulk $\text{Zn}_2(\text{benzimidazole})_4$

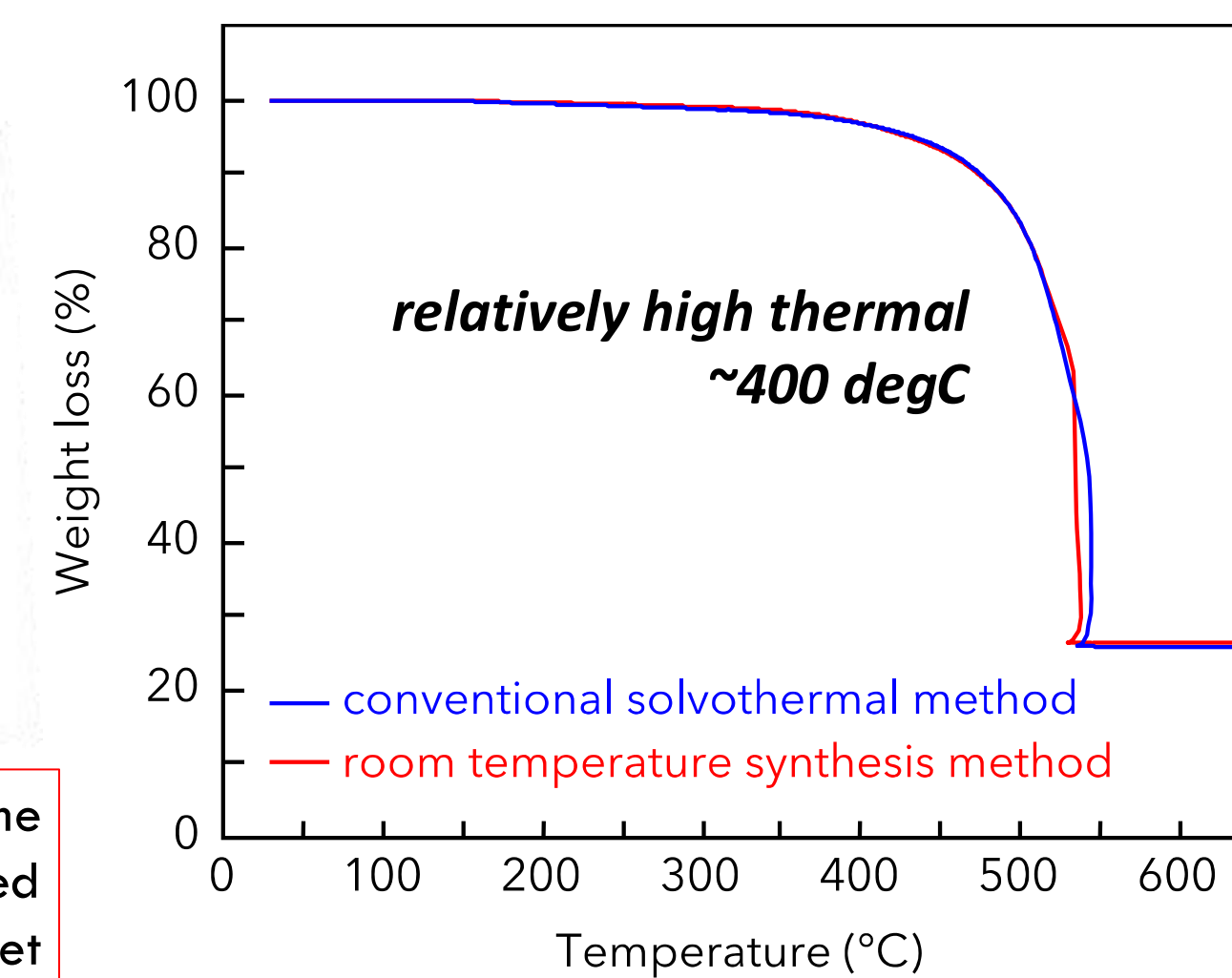


ZIF-7 sodalite topology, which comprises of interconnected six-membered rings with an inner cage size of 5.6 Å and pore aperture of 2.9 Å

layered structure

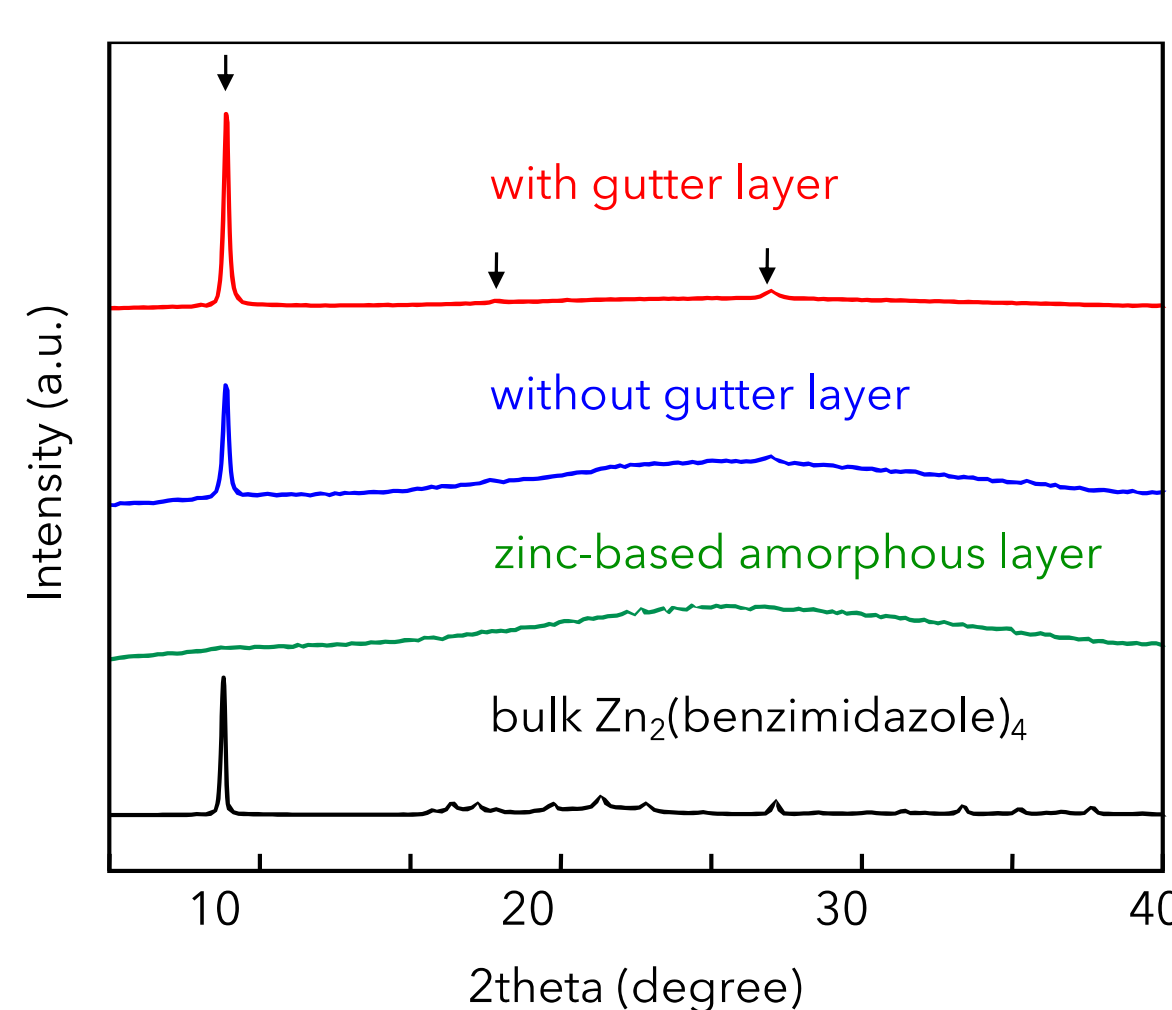


$\text{Zn}_2(\text{benzimidazole})_4$ with the same composition as ZIF-7 and layered structure, is a promising nanosheet precursor for hydrogen sieving



relatively high thermal ~400 degC

Conversion of zinc-based amorphous layer to $\text{Zn}_2(\text{benzimidazole})_4$



After Zn-based layer was exposed to benzimidazole vapor at 180 degC, sharp diffraction peaks appeared.

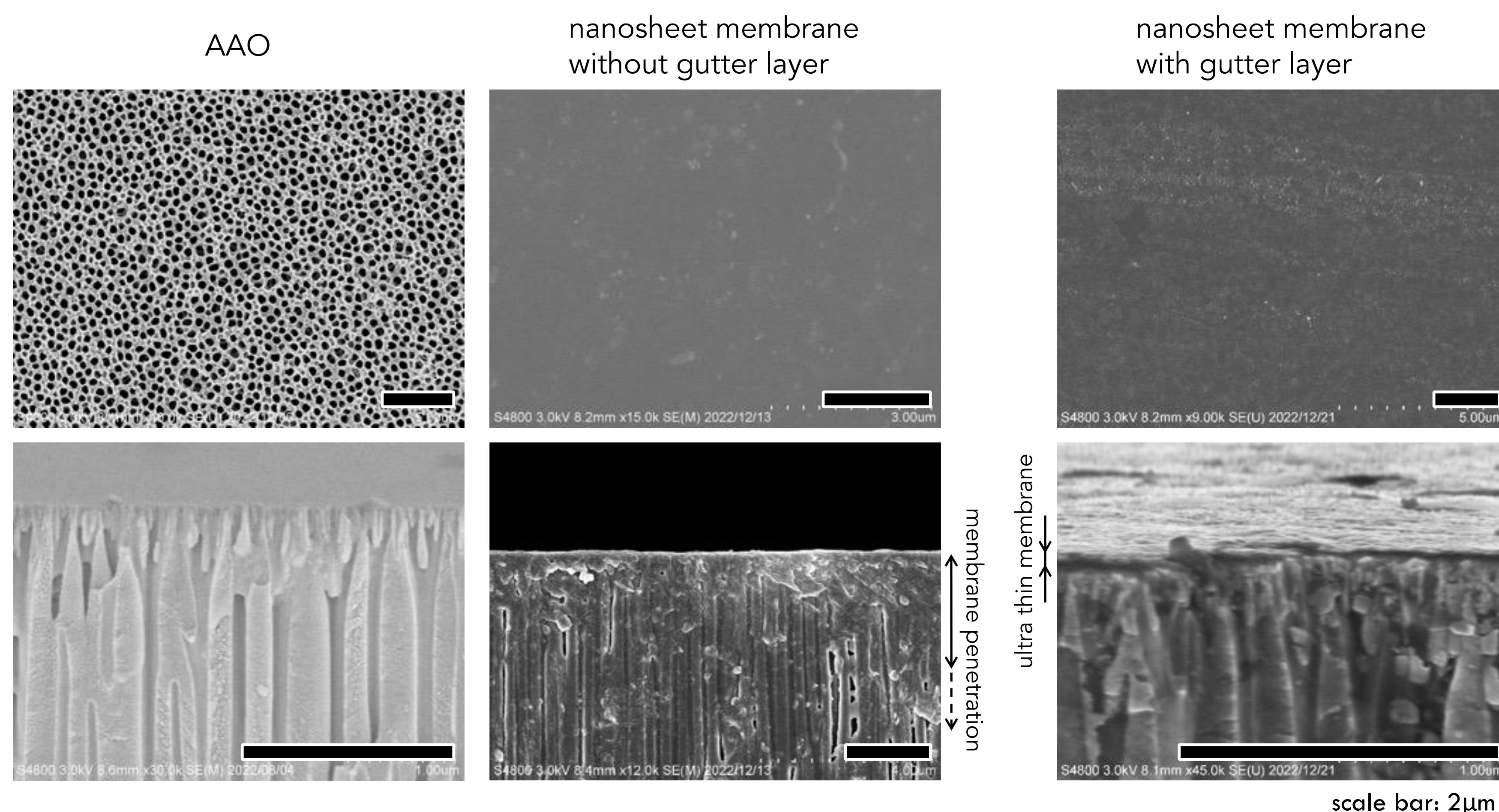
Zn-based layer shows broad diffraction.

The PXRD shows 1st, 2nd and 3rd order diffraction peaks, indicating that the structure of the layer is strongly oriented along the c axis.

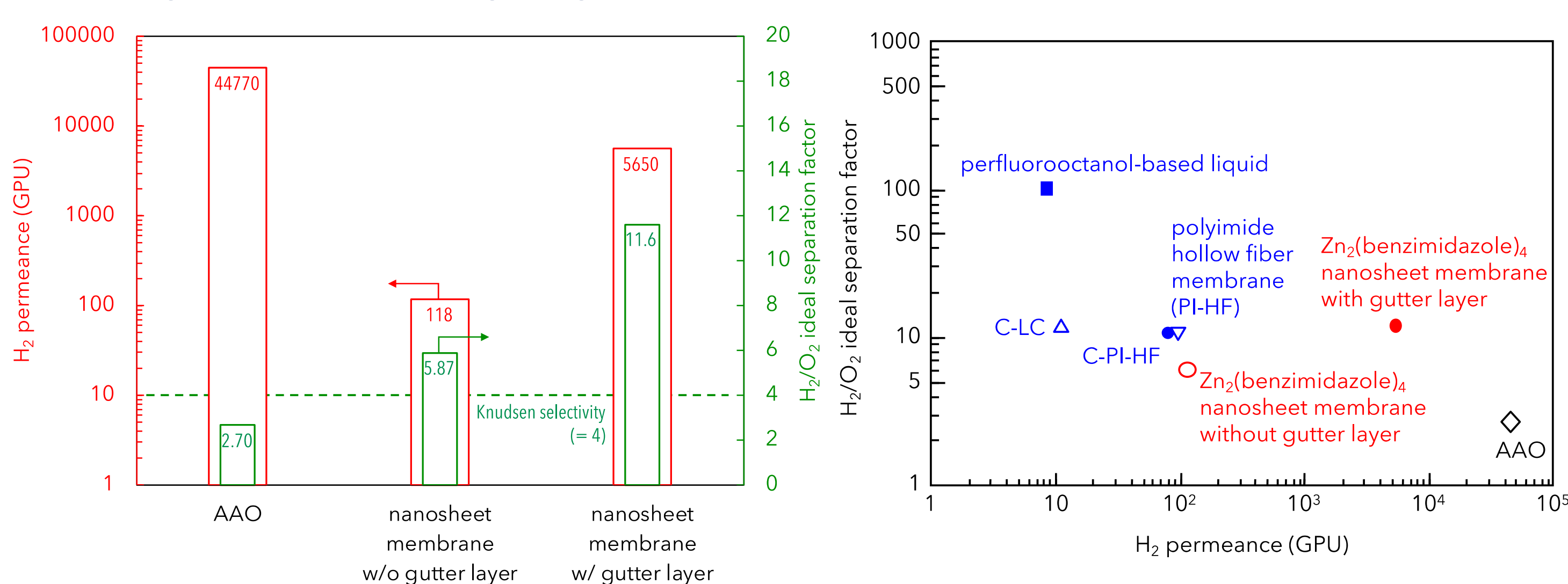
Crystallinity:

w/gutter layer \gg w/o gutter layer

When the $\text{Zn}_2(\text{benzimidazole})_4$ gutter layer was not coated on AAO, the broad diffraction around 25° still remained after benzimidazole vapor deposition, indicating that the membrane contains crystalline structure and amorphous region.

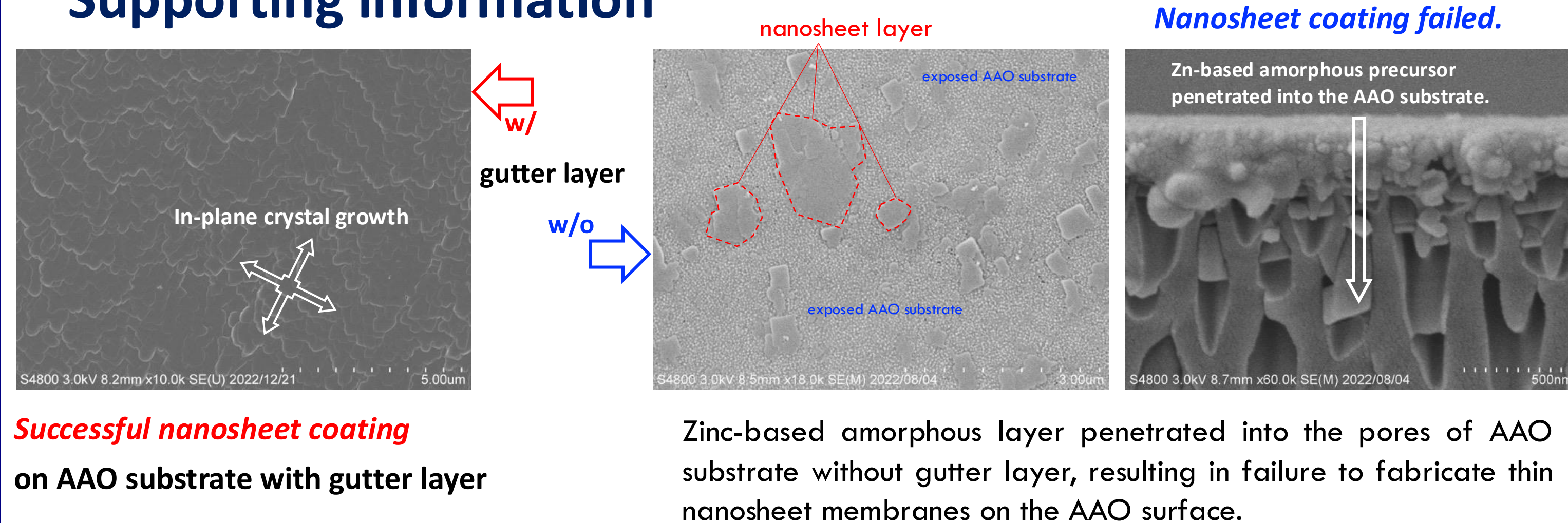


Gas permeation properties



The prepared $\text{Zn}_2(\text{benzimidazole})_4$ nanosheet-based membranes show separation performance in hydrogen purification with H_2/O_2 ideal separation factor of 11.6 and H_2 permeance of 5650 GPU.

Supporting information



Acknowledgements

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